West Side Science Club – Event *22 – "Battery Exploration"

Original Presentation

Date: 1 February 2014 Time: 10 am to 12 pm

Site: West Side Science Club

Big Questions

- These questions are meant to frame the day's event and might be written on the chalkboard
 - (1) What is a battery? How do they work?
 - (2) Are all types of batteries the same?

Concepts

 Concepts to cover from the "Work of CCI Solar" Mind Map: Reactions- oxidation/reduction, acid/base, yield, rate, starting materials, products; Social Aspects- climate change, greenhouse gases, global energy demand, fossil fuels, renewable vs nonrenewable energy; Energy- forms, storage, conversions.

Lesson Plan

Student Objectives

- Understand that not all batteries are created equally
- Be able to broadly define what constitutes a battery and how that are used
- Be introduced to renewable energy batteries

Schedule/Agenda

•	Review: Even	t "21–	· "H₂ Generation "			(10 min.)
					•	44.1.1

Activity: Comparison of different types of energy sources: (1 hr)

- Lemon Batteries
- Standard batteries
- o Fuel cells
- o Solar cells

•	Activity: Check out vintage car and its battery	(40 min.)
•	Wrap-up	(10 min.)

Materials

Activity: Battery comparisons

- Alkaline batteries
- Copper and zinc strips
- Lemons
- Alligator clips
- Multimeters
- Tiny devices to power batteries
- Fuel cells
- Solar cells

Activity: Car batteries

- A vintage car
- A modern car

Safety

• Though there are no serious risks involved in these experiments, safety glasses and gloves should be worn to promote good safety habits

Review of Previous Event: H2 Generation

· Recall the activities: H2 generation with Aluminum and Gallium; fuel cell electrolysis

Facilitation Questions

- What types of reactions were able to generate hydrogen?
- Why would we want to use hydrogen as a fuel?
- Do you think that hydrogen is a good fuel? How does it compare to other fuel sources we already use?

Activity: Comparison of different types of energy sources

Procedure

- 1. Using multimeters, test the amount of voltage and current that can come from the following sources
 - Lemon Batteries
 - Puncture a lemon at two points and insert a zinc electrode in one hole and a copper electrode in the other

- Then attach one end of an alligator clip to one electrode and the other end to the multimeter. Do the same with the opposite electrode
- Standard batteries
 - Attach the leads of the multimeter to each end of the battery
- Fuel cells
 - Attach an alligator clips to each output lead of the fuel cell and the other end of the clips to the multimeter
- Solar cells
 - Attach an alligator clips to each output lead of the solar cell and the other end of the clips to the multimeter
- 2. Multiply the current and voltage together to obtain the power output of each source
- 3. Next, based on power output results, try and power a variety of small devices such as LEDs, tiny motors and fans.

Facilitation/Concept Questions

- Which battery source do you expect to have the most power?
- Were your expectations correct?
- What changes do you think you could make to each battery source to make them output more power?
- What are the pros and cons of each power source?

Activity: Car Batteries

Procedure

- 1. Have students visually inspect the outside of a vintage car and make some predictions about how it works
- 2. Lift the hood and have the students now explore what the motor is like in an older car and how it runs
- 3. Next inspect the outside and under the hood of a modern car
- 4. Take note of the differences between the two cars and why certain changes have likely been made over time

Facilitation Questions and Concepts

- How are older cars powered? Newer cars?
- How can you integrate a battery into a car? Why has that improvement been made?
- How might a fuel cell be incorporated into a car?

Preparation for future week

Unveiling of the toy fuel cell car for the building competition in the next few weeks